

# Gas accumulations in Oligocene–Miocene reservoirs in the Alpine Foreland Basin (Austria): evidence for gas mixing and gas degradation

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**Abstract** Two petroleum systems are present in the eastern (Austrian) sector of the Alpine Foreland Basin. Whereas oil and thermogenic gas in Mesozoic and Eocene reservoir rocks have been generated beneath the Alps in Lower Oligocene source rocks, relative dry gas in Oligocene–Miocene clastic rocks deposited in the deep marine basin-axial channel system (Puchkirchen Channel) is interpreted as microbial in origin. Detailed investigations of the molecular and isotope composition of 87 gas samples from 86 wells, representing all producing fields with Oligocene and Miocene reservoir rocks, suggest that the presence of pure microbial gas is rare and limited mainly to the northern basin flank (e.g., KK field). All other fields contain varying amounts of thermogenic gas, which has been generated from a source rock with oil-window maturity. A relation with the underlying thermogenic petroleum system is obvious. Upward migration occurred along discrete fault zones (e.g., H field) or through low-permeability caprocks. Local erosion of Lower Oligocene sediments, the principal seal for the thermogenic petroleum system, as well as a high percentage of permeable rocks within the Puchkirchen Channel favored upward migration and mixing of thermogenic and microbial gas. All gas samples in Oligocene–Miocene reservoirs are biodegraded. Biodegradation and the formation of secondary microbial gas resulted in gas drying. Therefore, the gas samples analyzed in this study

are relative dry, despite significant contributions of thermogenic hydrocarbons. Biodegradation probably continues at present time. The degree of biodegradation, however, decreases with depth.

**Keywords** Foreland basin · Gas mixing · Gas biodegradation · Secondary microbial methane

## Introduction

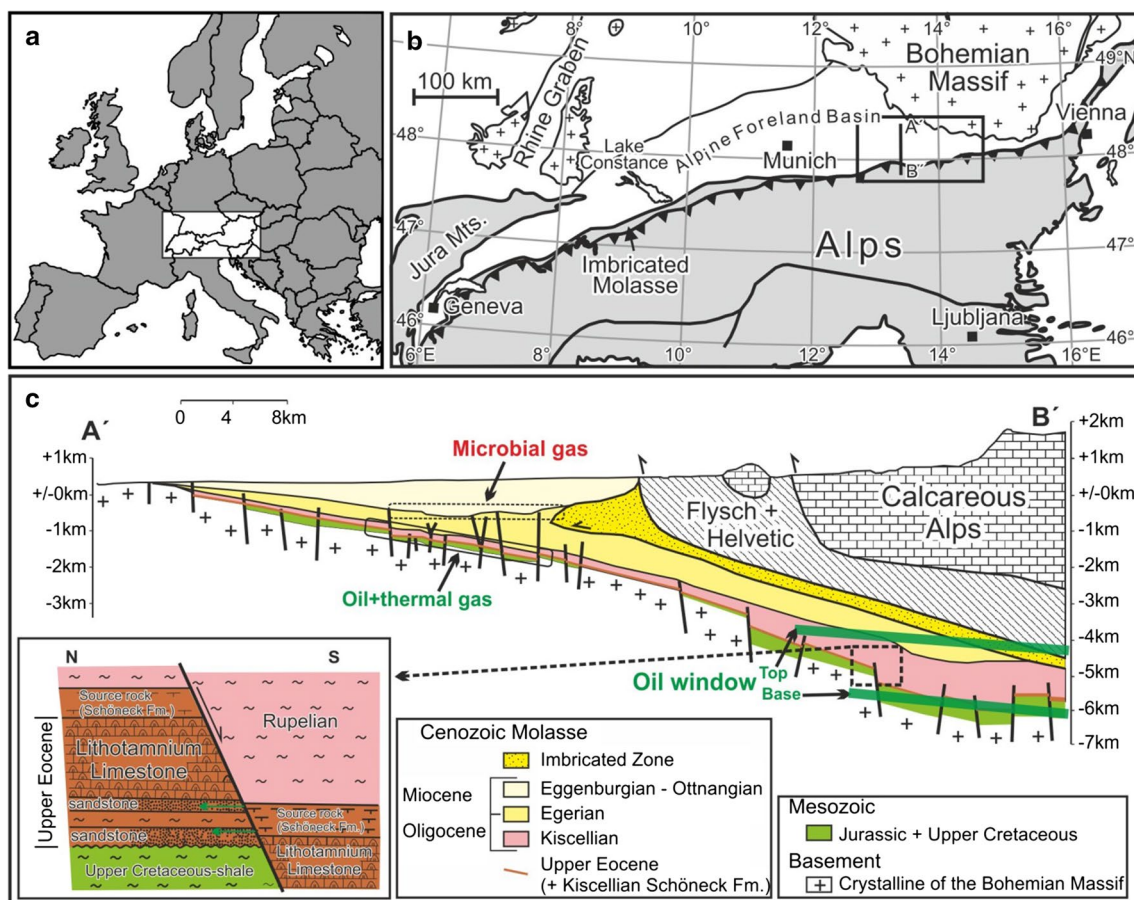
Significant progress has been achieved in the last decades in the understanding of the formation, mixing and alteration of natural gas as well as of gas-oil and gas-source rock correlations. Based on the molecular and isotope composition of gas, it has been recognized that in addition to thermogenic processes and biodegradation of organic matter in pelitic rocks (primary microbial gas), commercial gas accumulation may result from biodegradation of preexisting oil and gas (secondary microbial gas) (e.g., Head et al. 2003; Jones et al. 2008; Milkov 2010, 2011).

Important parameters for the interpretation of gas data are stable carbon and hydrogen isotope ratios. Stable carbon isotopic ratios are mainly controlled by the formation mechanism (thermogenic vs. microbial), as well as the isotopic signature and thermal maturity of the source rocks (Berner and Faber 1987; Chung et al. 1988; Clayton 1991; Fuex 1977; Galimov 2006; James 1983, 1990; Krooss et al. 1995; Littke et al. 1995; Rice and Claypool 1981; Rooney et al. 1995; Schoell 1980, 1983; Stahl 1977; Whiticar 1994). Hydrogen isotope ratios are controlled by the exchange of hydrogen between water and thermally maturing organic matter (Lewan 1993, 1997; Schimmelmann et al. 2001, 2004; Yoneyama et al. 2002). In contrast, the exchange between water and already formed *n*-alkanes

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**Fig. 1** a, b Location maps of study area, c cross section through the Alpine Foreland Basin (modified after Wagner 1996). Inset in c explains migration from source rocks into stratigraphically deeper carrier beds across normal faults (Malzer et al. 1993). See b for location of section

is limited (Hoering 1984; Sessions et al. 2004; Seewald et al. 1998). Enrichment in deuterium with increasing thermal stress is observed in natural samples (Dai 1990; Radke et al. 2005; Schoell 1980) as well as in theoretical and laboratory studies (e.g., Ni et al. 2011; Tang et al. 2005).

Since the pioneering work on carbon and hydrogen isotope ratios of Schoell (1977, 1984), it is known that both oil-associated thermogenic gas and microbial methane exist in the Alpine Foreland Basin, which extends along the northern margin of the Alps from Vienna to Geneva (Fig. 1a, b). According to traditional concepts, thermogenic hydrocarbons are found in Mesozoic and Eocene reservoirs, whereas microbial gas prevails in Oligocene–Miocene reservoirs separated effectively from the underlying thermogenic petroleum system by several hundred meters of shales (e.g., Brix and Schultz 1993).

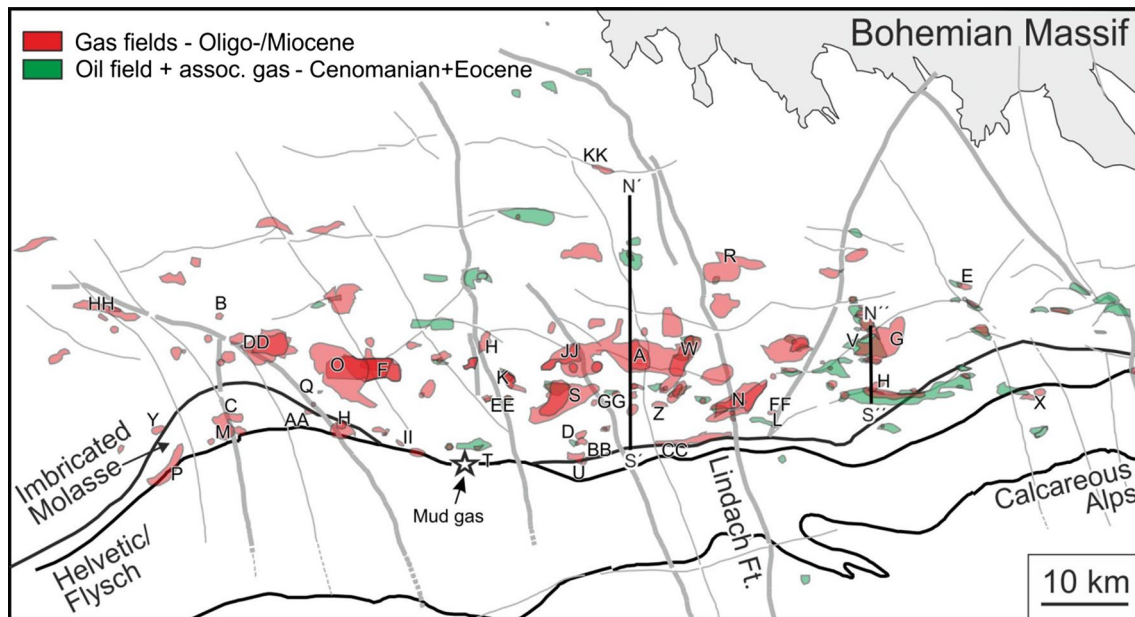
However, recently Reischenbacher and Sachsenhofer (2011) emphasized that molecular and isotopic gas data from the Austrian part of the Alpine Foreland Basin are in conflict with this simple model. Actually, Pytlak et al. (2016) could show that methane associated with oil

deposits in Cenomanian and Eocene reservoirs is a mixture of thermogenic, primary and secondary microbial methane. Moreover, the presence of condensate in some Oligocene–Miocene reservoirs challenges the interpretation of pure microbial gas in Oligocene–Miocene reservoirs.

The present paper focusses on gas in Upper Oligocene and Lower Miocene clastic reservoirs. Main aims of the present contribution are to determine the origin of the gas and to understand possible mixing and alteration processes. To reach this goal, gas samples from 86 wells representing all producing fields in the Austrian sector of the basin (fields labeled in Fig. 2) were studied for their molecular and isotopic composition. Note that the present paper deals with gaseous hydrocarbons and condensates will be studied in a separate paper.

## Geological background

The Alpine Foreland Basin extends along the northern margin of the Alps between Geneva and Vienna (Fig. 1a,



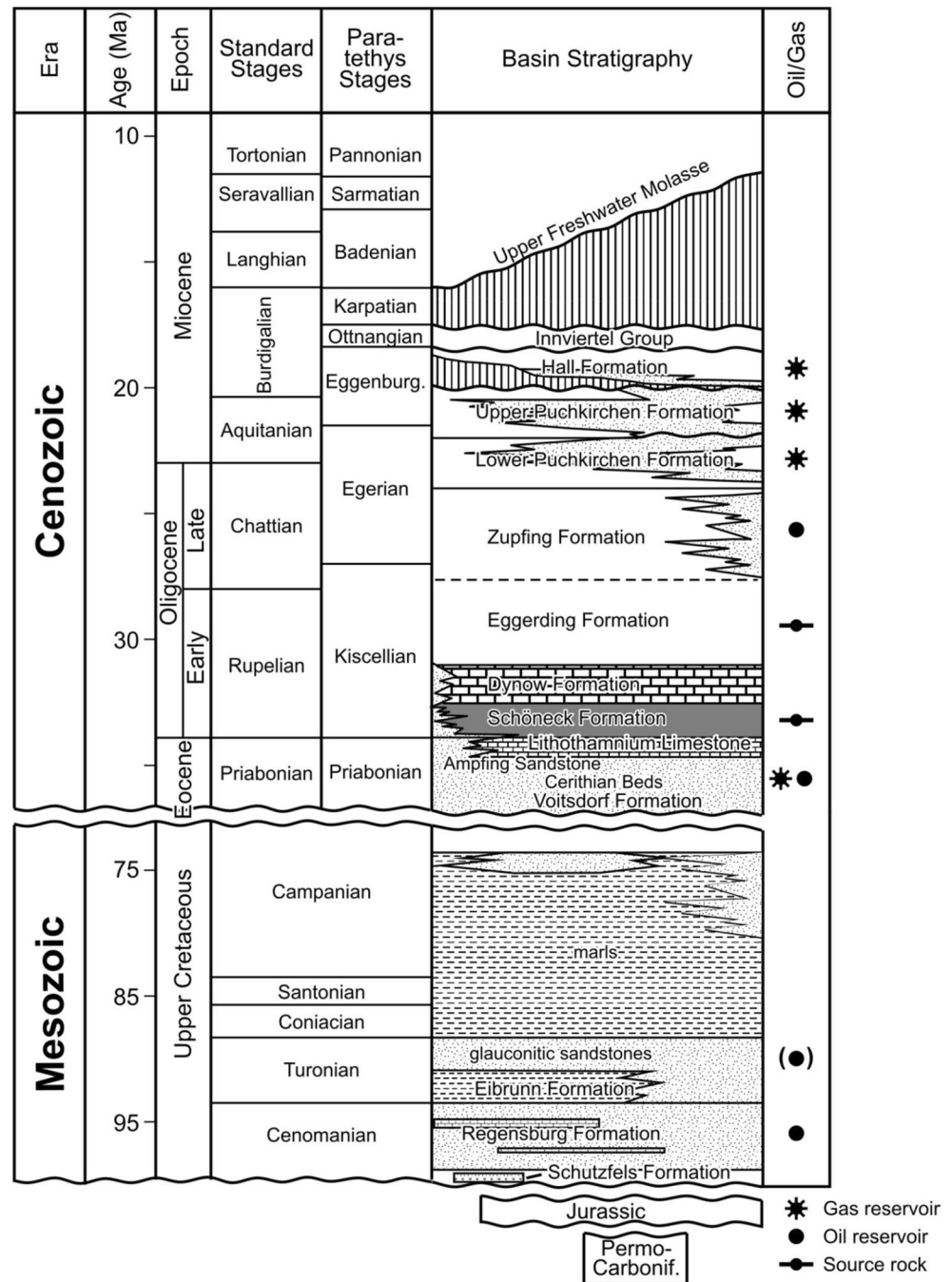
**Fig. 2** Oil and gas fields in study area. Labels denote currently producing gas fields with Oligocene–Miocene reservoirs. Cross sections N'–S' and N''–S'' are presented in Fig. 10a, b

b). The asymmetric basin deepens toward the south and extends below the Alpine nappes (Fig. 1c). In the Austrian sector of the Alpine Foreland Basin, the sedimentary succession overlies crystalline basement of the Bohemian Massif and comprises from bottom to top: (1) Permo-Carboniferous graben sediments, (2) Jurassic and Upper Cretaceous mixed carbonate-siliciclastic shelf sediments, and (3) Eocene to Upper Miocene Molasse sediments (Fig. 3). The Cenozoic section starts with Upper Eocene shallow-marine and fluvial carbonates, shales and sandstones. Late Eocene and Early Oligocene loading of the Alpine nappes caused down-bending of the foreland, southward increasing basement depth and rapid deepening of the depositional environment. Deep marine pelitic rocks with thin carbonate layers were deposited during Early Oligocene time (Schöneck, Dynow, Eggerding formations). Locally these rocks, about 60 m thick, have been removed by submarine erosion before deposition of the overlying Zupfing Formation. During Late Oligocene and Early Miocene times gravity flow deposits (Puchkirchen Group, lower Hall Formation; Fig. 3) accumulated in a deep marine basin-axial channel system (Puchkirchen Channel) originating in the west (near Munich) from a prograding–retrograding delta system (Covault et al. 2009; Hubbard et al. 2005, 2009). The coarse-grained clastic rocks form part of the Lower Puchkirchen Formation (uppermost Chattian to lowermost Aquitanian), the Upper Puchkirchen Formation (Aquitanian to lower Burdigalian) and the Hall Formation, which follow above a major erosional event (Grunert et al. 2013, 2015). Pelitic rocks, typically with low organic matter

contents, were deposited outside of the channel system. However, organic matter-rich intervals, including a “fish shale” near the top of the Upper Puchkirchen Formation, occur as well (Wagner 1996, 1998). Decreased subsidence and increased sedimentation rates resulted in filling of the basin between late Eggenburgian and Otnangian times. Coal-bearing freshwater sediments with a Middle to Late Miocene age dominate the top of the Cenozoic succession. Different fault systems occur in the Alpine Foreland Basin. NE–SW and NW–SE trending faults formed during Paleozoic times and were reactivated in the Mesozoic and the Paleogene. During Late Eocene and Early Oligocene times flexural down-bending resulted in the formation of W–E trending normal faults. Many of them were reactivated during the Miocene (Wagner 1996, 1998). Extensive regional uplift after Late Miocene time caused erosion of sediments, 500–900 m thick and cooling of about 20 °C (Gusterhuber et al. 2012).

Traditionally, two petroleum systems are distinguished in the Austrian part of the Alpine Foreland Basin: A thermogenic petroleum system is based on Lower Oligocene source rocks (Schöneck Fm., Eggerding Fm.; Sachsenhofer et al. 2010; Sachsenhofer and Schulz 2006; Schulz et al. 2002), which reached the oil window beneath the Alpine nappes in Miocene time (Gusterhuber et al. 2013). Main reservoirs are Cenomanian and Upper Eocene non-marine to shallow-marine sandstones in structural traps. The Lower Oligocene Schöneck Formation is also the primary seal formation (Veron 2005). The cumulative production of oil + condensate was 8903825 tons and 1647 M Nm<sup>3</sup>

**Fig. 3** Stratigraphy of Cenozoic and Mesozoic rocks in the Austrian part of the Alpine Foreland Basin (after Wagner 1996)



of associated gas (RAG production until 2015, industrial data).

Methane depleted in  $^{13}\text{C}$ , traditionally interpreted as microbial in origin (e.g., Schoell 1984; Schulz and van Berk 2009), prevails in clastic deep water sediments with a Late Oligocene to Early Miocene age (Puchkirchen Group, Hall Fm.; Covault et al. 2009; Hubbard et al. 2005, 2009). Intraformational shales form the seals (Veron 2005). The occurrence of condensate and wet gas suggests at least local mixing with thermogenic hydrocarbons (Reischenbacher and Sachsenhofer 2011; Wagner 1998). The

cumulative production from Oligocene and Miocene units was 23352 M Nm<sup>3</sup> of gas (RAG production until 2015, industrial data).

## Sampling and analytical methods

### Sampling procedure

Samples (87) have been taken from 86 wells using the Iso-Tube<sup>®</sup> gas sampling system. The aluminum-made sampling



cylinders were attached directly to the wellheads via the pressure reduction unit and purged with produced gas several times to avoid air contamination. Finally, the IsoTubes were filled to the maximum allowed pressure of 8 bar. In addition, mud gas data of a recently drilled well are considered in the interpretation section.

### Laboratory methods

Molecular compositions of gas samples were determined by mounting the IsoTubes directly to the sampling loop of a Trace GC-ultra equipped with three various gas channels. Hydrocarbons were resolved on a 30-m Rtx-Alumina capillary column (i.d. 0.53 mm; filling  $\text{Na}_2\text{SO}_4$ , 10  $\mu\text{m}$  film thickness) and detected by FID. Permanent gases were resolved on two packed columns: HayeSep Q (2 m  $\times$  1/8" OD) and MolSieve 5A (2 m  $\times$  1/8" OD) and detected by TCD. The column oven was programmed to hold a temperature of 50 °C for 4.3 min and then heated to 165 °C at 10 °C/min, at which point the temperature was held for 0.5 min. Helium was used as carrier gas for all three channels. Peak identification has been performed based on comparison of retention times of calibration gas mixtures and analyzed samples. Concentrations have been calculated based on peak areas against calibration gas mixtures consisting of most compounds observed in the samples.

Stable C and H isotope measurements were carried out using a Trace GC-ultra gas chromatograph attached to the ThermoFisher Delta-V isotope ratio mass spectrometer (IRMS) via a combustion and high-temperature reduction interface, respectively (GC Isolink, ThermoFisher). The GC coupled to the IRMS was equipped with a 25-m PoraPlot capillary column (i.d. 0.32 mm; 0.10  $\mu\text{m}$  film thickness). The oven temperature was programmed from 30 to 180 °C at a rate of 5 °C/min followed by an isothermal period of 5 min. Helium was used as carrier gas. For calibration, a  $\text{CO}_2$  or  $\text{H}_2$  standard gas was injected at the beginning and at the end of each analysis. Analytical reproducibility was controlled by repeated measurements of a calibration gas.  $^2\text{H}/^1\text{H}$  and  $^{13}\text{C}/^{12}\text{C}$  relative ratios are expressed relative to VSMOW and VPDB, respectively.

## Results

### Molecular composition

The studied gas samples are dominated by methane (97.2–99.4%) with minor concentrations of  $\text{C}_{2+}$  hydrocarbons, nitrogen (0.2–2.1%) and carbon dioxide (<0.5%). In few samples traces of ethene and/or propene were detected (Table 1). Traces of different  $\text{C}_{5+}$  isomers have been detected, but only some of them have been quantified

(Fig. 4). Thus, results reported in Table 1 were not normalized to 100%.

Gas dryness, defined by  $\text{C}_1/(\text{C}_2 + \text{C}_3)$  (Bernard et al. 1978), varies between 96 and 471 with an average value of 228. An exceptional dry gas (~4400) is produced from the KK field (Fig. 5) in the northern part of the basin (see Fig. 2 for location).

Ratios of branched versus straight *n*-alkanes range between 0.9 and 4.4 for butanes and 1.9 and 19.8 for pentanes, respectively.

### Isotopic composition

The stable carbon isotopic composition of methane varies widely from  $-48.5$  to  $-65.1\text{‰}$ . Methane, most enriched in  $^{13}\text{C}$  has been found in the F–O fields located in the western part of study area (Table 1; Fig. 5), whereas methane, most depleted in  $^{13}\text{C}$  is from the KK field.  $\delta^2\text{H}$  of methane ranges between  $-178$  and  $-221\text{‰}$  (Fig. 6). Similarly to  $\delta^{13}\text{C}$  distribution, methane with elevated  $^2\text{H}$  content occurs in the F–O fields. In addition, methane enriched in  $^2\text{H}$  is recorded in gas from the X field (Table 1).

$\text{C}_2$  ( $-55.7$  to  $-33\text{‰}$ ),  $\text{C}_3$  ( $-38.5$  to  $-24.5\text{‰}$ ) to *n*- $\text{C}_4$  ( $-33.4$  to  $-17\text{‰}$ ) hydrocarbons show a decreasing variability in stable carbon isotope ratios (Table 1).

Stable isotope ratios of methane obtained in the present study show a wider range of values than carbon ( $-59.7$  to  $-63.3\text{‰}$ ) and hydrogen isotope ratios ( $-197$  to  $-212\text{‰}$ ) reported by Schoell (1984) and Schulz and van Berk (2009). This is because these authors coincidentally did not sample wells with isotopically heavy methane.

## Interpretation and discussion

### Microbial versus thermogenic gas

The driest gas is present in the KK field, where only methane and traces of ethane are observed. Moreover, methane from this field is characterized by the most negative carbon isotopic values (Fig. 5; Table 1). This suggests a pure microbial origin of the KK gas, located in the shallow northern flank of the basin, and confirms the established models (e.g., Brix and Schultz 1993; Schoell 1984). Interestingly, ethane from KK gas is isotopically lighter than that of oil-associated gas from Mesozoic and Eocene reservoirs (Fig. 7, Pytlak et al. 2016).

All other gas samples contain hydrocarbons up to *n*-hexane and yield significantly lower dryness values. Moreover, a significant number of them display isotope values characteristic for varying contributions of thermogenic hydrocarbons (Figs. 5, 6). Carbon isotope values as low as  $-48.5\text{‰}$  (Fig. 5) indicate the presence of the highest amount of

**Table 1** Molecular and isotopic composition of sampled gas

Well	TVDS	Formation	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	i-C <sub>4</sub> H <sub>10</sub>	n-C <sub>4</sub> H <sub>10</sub>	i-C <sub>3</sub> H <sub>12</sub>	n-C <sub>3</sub> H <sub>12</sub>	n-C <sub>6</sub> H <sub>14</sub> +	CO <sub>2</sub>	N <sub>2</sub>	δ <sup>13</sup> C <sub>CH<sub>4</sub></sub>	δ <sup>13</sup> C <sub>C<sub>2</sub>H<sub>6</sub></sub>	δ <sup>13</sup> C <sub>C<sub>3</sub>H<sub>8</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>i-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>CO<sub>2</sub></sub>	δ <sup>2</sup> H <sub>CH<sub>4</sub></sub>
A-1	1171	Upper PF.	98.55	0.404	0.124	0.046	0.019	0.025	0.006	0.002	0.213	0.489	-59.0	-47.2	-33.8	-25.2	-26.0	-6.0	-205
A-2	606	H.F.	98.34	0.338	0.114	0.045	0.014	0.019	0.004	0.000	0.008	0.482	-62.5	-41.9	-29.3	-22.7	-23.2	-8.2	-198
A-3	1182	Upper PF.	97.99	0.416	0.123	0.043	0.020	0.026	0.006	0.002	0.073	0.497	-58.0	-48.8	-35.0	-26.0	-26.5	-5.9	-207
A-4	1165	Upper PF.	98.20	0.482	0.202	0.081	0.029	0.041	0.009	0.002	0.131	0.602	-59.5	-45.7	-34.5	-27.5	-27.9	-5.7	-205
A-5	1169	Upper PF.	98.37	0.424	0.147	0.049	0.021	0.027	0.007	0.002	0.160	0.507	-59.0	-47.0	-35.0	-26.3	-26.4	-5.7	-206
A-6	1167	Upper PF.	98.57	0.405	0.131	0.045	0.024	0.027	0.007	0.003	0.106	0.511	-59.8	-45.9	-35.3	-25.8	-24.5	-6.7	-205
A-7	1399	Upper PF.	97.86	0.545	0.220	0.103	0.031	0.041	0.008	0.002	0.103	0.674	-57.8	-45.8	-34.6	-31.1	-30.3	-5.7	-202
B-1	802	H.F.	98.29	0.349	0.162	0.046	0.018	0.020	0.005	0.001	0.020	0.617	-58.4	-48.6	-33.0	-21.9	-22.0	18.0	-210
C-1	749	Upper PF.	98.40	0.206	0.052	0.017	0.005	0.007	0.002	0.001	0.009	0.393	-62.0	-55.7	-36.0	-25.0	-24.2	-6.5	-221
D-1	1021	Upper PF.	98.19	0.358	0.106	0.032	0.018	0.021	0.005	0.001	0.041	0.430	-57.6	-49.2	-24.7	-26.2	-22.0	5.0	-204
E-1	33	H.F.	98.92	0.277	0.081	0.022	0.007	0.006	0.001	0.000	0.035	0.653	-62.7	-38.0	-26.8	-24.0	-22.0	-38.0	-204
F-1	1093	Upper PF.	98.61	0.332	0.160	0.058	0.048	0.033	0.016	0.004	0.076	0.777	-61.4	-42.8	-31.0	-26.0	-24.7	-4.0	-201
F-2	690	H.F.	98.80	0.327	0.124	0.044	0.018	0.020	0.005	0.001	0.072	0.560	-60.5	-46.3	-31.9	-26.0	-24.2	3.0	-202
F-3	1694	Upper PF.	97.68	0.687	0.335	0.126	0.064	0.072	0.016	0.004	0.298	0.766	-59.6	-43.5	-35.5	-30.9	-30.2	-1.3	-203
F-4	1061	Upper PF.	98.28	0.356	0.106	0.033	0.022	0.018	0.006	0.002	0.069	0.516	-48.5	-38.0	-29.4	-27.2	-25.4	0.1	-185
F-5	702	H.F.	98.40	0.363	0.108	0.033	0.020	0.023	0.007	0.002	0.098	0.459	-56.1	-43.2	-33.5	-28.8	-23.5	-1.4	-180
F-6	691	H.F.	98.54	0.327	0.126	0.044	0.019	0.019	0.005	0.001	0.007	0.616	-50.3	-41.1	-31.6	-29.4	-26.3	-0.3	-193
G-1	234	H.F.	98.94	0.304	0.100	0.029	0.012	0.013	0.003	0.000	0.062	0.516	-63.5	-38.5	-25.7	-25.8	-24.7	-7.1	-202
H-1	305	H.F.	98.65	0.253	0.074	0.018	0.006	0.007	0.002	0.001	0.057	0.471	-61.6	-36.4	-24.5	-17.0	-17.8	-3.8	-199
I-1	627	H.F.	98.73	0.330	0.113	0.035	0.014	0.015	0.004	0.001	0.063	0.420	-60.0	-49.6	-32.0	-26.0	-25.2	4.5	-201
I-2	695	Upper PF.	98.82	0.272	0.082	0.019	0.010	0.009	0.003	0.001	0.030	0.373	-59.3	-50.8	-34.1	-26.0	-25.0	4.0	-211
I-3	689	Upper PF.	98.42	0.263	0.075	0.019	0.010	0.010	0.003	0.002	0.036	0.376	-58.9	-51.3	-33.2	-25.5	-22.0	-7.4	-204
J-1	1410	Upper PF.	97.24	0.484	0.221	0.071	0.027	0.043	0.009	0.004	0.261	0.725	-61.6	-38.8	-35.7	-25.2	-23.6	-7.4	-204
K-1	1151	Upper PF.	98.29	0.371	0.113	0.033	0.017	0.022	0.006	0.002	0.407	0.422	-57.4	-50.8	-33.7	-28.0	-27.7	-7.0	-205
L-1	398	H.F.	99.41	0.273	0.125	0.029	0.011	0.012	0.003	0.005	0.046	0.464	-60.2	-45.1	-30.5	-25.3	-23.5	-7.9	-206
M-1	714	Upper PF.	98.46	0.169	0.067	0.016	0.007	0.006	0.001	0.000	0.029	0.447	-60.7	-51.6	-34.2	-31.9	-30.8	-10.0	-212
M-2	629	Upper PF.	98.71	0.311	0.065	0.013	0.005	0.005	0.002	0.000	0.114	0.493	-60.9	-54.1	-33.6	-26.0	-23.4	-16.0	-216
M-3	742	Upper PF.	98.77	0.267	0.052	0.013	0.005	0.006	0.002	0.001	0.064	0.480	-61.0	-53.4	-35.6	-26.0	-24.6	-17.0	-218
M-4	763	Upper PF.	98.92	0.254	0.070	0.016	0.007	0.007	0.002	0.000	0.038	0.407	-62.1	-52.2	-34.7	-33.4	-29.9	-9.1	-202
M-5	571	Upper PF.	98.39	0.226	0.054	0.014	0.008	0.006	0.002	0.000	0.085	0.391	-60.3	-52.2	-33.9	-30.9	-29.3	-10.8	-205
M-6	577	Upper PF.	97.75	0.262	0.049	0.012	0.005	0.006	0.002	0.001	0.119	0.468	-60.4	-46.0	-34.0	-26.0	-22.4	-15.5	-221
N-1	446	H.F.	98.33	0.318	0.102	0.036	0.010	0.013	0.001	0.003	0.166	0.401	-60.0	-47.5	-28.7	-27.0	-24.0	-8.6	-203
N-2	335	H.F.	99.05	0.317	0.097	0.019	0.006	0.008	0.000	0.000	0.026	0.480	-60.5	-47.2	-31.0	-25.5	-22.0	-8.9	-202
N-3	360	H.F.	98.83	0.302	0.090	0.019	0.006	0.018	0.001	0.000	0.038	0.693	-60.9	-49.6	-30.4	-25.5	-25.0	-4.0	-200
O-1	690	H.F.	97.97	0.319	0.127	0.032	0.011	0.014	0.003	0.000	0.154	0.471	-59.7	-49.9	-33.6	-31.0	-29.9	-9.2	-189

Table 1 continued

Well	TVDS	Formation	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	i-C <sub>4</sub> H <sub>10</sub>	n-C <sub>4</sub> H <sub>10</sub>	i-C <sub>5</sub> H <sub>12</sub>	n-C <sub>5</sub> H <sub>12</sub>	n-C <sub>6</sub> H <sub>14</sub> +	CO <sub>2</sub>	N <sub>2</sub>	δ <sup>13</sup> C <sub>CH<sub>4</sub></sub>	δ <sup>13</sup> C <sub>C<sub>2</sub>H<sub>6</sub></sub>	δ <sup>13</sup> C <sub>C<sub>3</sub>H<sub>8</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>i-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>5</sub>H<sub>12</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>6</sub>H<sub>14</sub>+</sub>	δ <sup>13</sup> C <sub>CO<sub>2</sub></sub>	δ <sup>13</sup> C <sub>CH<sub>4</sub></sub>
O-2	673	H.F.	98.00	0.319	0.099	0.036	0.015	0.015	0.003	0.001	0.102	0.566	-55.3	-39.4	-30.6	-27.8	-26.7	-26.7	-26.7	-6.4	-187
O-3	744	H.F.	97.84	0.326	0.101	0.032	0.013	0.014	0.003	0.001	0.034	0.485	-49.6	-40.8	-31.8	-29.5	-26.0	-26.0	-26.0	-6.9	-200
O-4	776	H.F.	98.47	0.427	0.144	0.054	0.022	0.027	0.006	0.002	0.127	0.506	-54.3	-43.3	-32.1	-28.3	-27.2	-27.2	-27.2	-8.2	-189
O-5	700	H.F.	98.76	0.326	0.116	0.034	0.011	0.013	0.002	0.000	0.024	0.480	-52.5	-43.4	-33.9	-28.9	-26.4	-26.4	-26.4	-10.8	-205
O-6	698	H.F.	97.89	0.328	0.123	0.034	0.014	0.015	0.004	0.001	0.054	0.488	-53.7	-42.1	-29.9	-27.8	-25.7	-25.7	-25.7	-8.2	-192
O-7	679	H.F.	97.56	0.317	0.120	0.030	0.009	0.011	0.002	0.000	0.082	0.580	-55.0	-44.9	-31.3	-28.7	-27.3	-27.3	-27.3	-8.0	-197
P-1	797	Upper P.F.	98.47	0.281	0.040	0.011	0.004	0.005	0.002	0.002	0.072	0.241	-54.5	-42.9	-31.2	-29.3	-28.5	-28.5	-28.5	1.8	-187
P-2	1045	Upper P.F.	98.77	0.282	0.042	0.005	0.003	0.002	0.001	0.000	0.052	0.219	-64.4	-55.1	-31.6	-29.3	-27.9	-27.9	-27.9	-7.1	-178
Q-1	1610	Upper P.F.	98.45	0.408	0.140	0.046	0.018	0.018	0.003	0.000	0.004	0.578	-60.3	-47.9	-34.7	-27.8	-28.0	-28.0	-28.0	-20.7	-207
R-1	273	H.F.	97.96	0.326	0.083	0.023	0.013	0.011	0.005	0.004	0.035	0.558	-61.3	-42.7	-28.8	-23.2	-22.0	-22.0	-22.0	-6.5	-202
R-2	254	H.F.	97.99	0.325	0.077	0.021	0.010	0.008	0.002	0.000	0.021	0.561	-61.6	-42.9	-28.5	-21.8	-20.2	-20.2	-20.2	-7.7	-202
R-3	317	H.F.	97.56	0.329	0.080	0.025	0.011	0.011	0.002	0.000	0.000	0.504	-61.3	-43.0	-28.4	-20.6	-22.3	-22.3	-22.3	-20.2	-202
R-4	275	H.F.	98.74	0.322	0.088	0.025	0.011	0.012	0.003	0.000	0.016	0.613	-62.6	-43.5	-29.5	-23.1	-23.5	-23.5	-23.5	-7.0	-202
R-5	284	H.F.	97.33	0.293	0.080	0.022	0.010	0.010	0.002	0.000	0.024	0.596	-63.0	-43.2	-28.4	-18.3	-21.1	-21.1	-21.1	-3.0	-201
S-1	447	H.F.	98.95	0.295	0.121	0.028	0.010	0.010	0.002	0.000	0.018	0.557	-60.7	-44.5	-28.8	-21.0	-21.0	-21.0	-21.0	-5.0	-203
S-2	1184	Upper P.F.	98.40	0.371	0.146	0.043	0.023	0.024	0.008	0.003	0.252	0.446	-58.1	-48.1	-34.6	-27.5	-27.5	-27.5	-27.5	-3.4	-207
S-3	1180	Upper P.F.	97.70	0.363	0.142	0.040	0.019	0.026	0.006	0.000	0.075	0.438	-57.8	-48.9	-34.2	-23.8	-24.4	-24.4	-24.4	4.2	-206
S-4	604	H.F.	98.99	0.333	0.110	0.028	0.018	0.018	0.006	0.000	0.038	0.439	-58.2	-51.0	-34.0	-28.0	-25.0	-25.0	-25.0	-11.0	-205
T-1	795	Upper P.F.	98.05	0.324	0.109	0.028	0.013	0.016	0.004	0.001	0.045	0.455	-63.8	-49.0	-27.3	-25.3	-22.8	-22.8	-22.8	0.0	-190
U-1	814	Upper P.F.	98.19	0.339	0.165	0.036	0.025	0.021	0.007	0.001	0.021	0.412	-59.3	-49.2	-34.0	-26.8	-26.4	-26.4	-26.4	-6.5	-204
U-2	739	Upper P.F.	98.48	0.334	0.161	0.036	0.026	0.022	0.008	0.002	0.064	0.399	-60.0	-46.5	-31.3	-25.0	-23.4	-23.4	-23.4	-7.3	-203
U-3	1037	Upper P.F.	98.44	0.363	0.222	0.044	0.034	0.030	0.009	0.002	0.044	0.725	-61.8	-49.7	-33.9	-25.7	-26.0	-26.0	-26.0	-6.0	-197
V-1	375	H.F.	98.88	0.260	0.087	0.024	0.007	0.010	0.002	0.000	0.054	0.498	-60.2	-33.0	-26.0	-25.8	-25.7	-25.7	-25.7	-6.0	-209
V-2	228	H.F.	98.82	0.323	0.148	0.037	0.029	0.019	0.010	0.008	0.055	0.547	-62.1	-36.0	-27.3	-27.2	-27.4	-27.4	-27.4	-8.0	-201
W-1	489	H.F.	97.46	0.503	0.195	0.086	0.026	0.038	0.007	0.002	0.068	0.633	-57.9	-45.8	-34.2	-25.8	-28.9	-28.9	-28.9	-9.0	-202
W-2	498	H.F.	97.85	0.472	0.151	0.071	0.029	0.040	0.010	0.003	0.394	0.511	-56.0	-46.0	-34.8	-28.9	-29.2	-29.2	-29.2	-2.6	-208
W-3	1156	Upper P.F.	97.61	0.537	0.209	0.096	0.029	0.037	0.007	0.002	0.131	0.656	-57.6	-45.4	-34.6	-28.4	-28.7	-28.7	-28.7	-7.8	-202
W-4	1162	Upper P.F.	97.99	0.448	0.128	0.051	0.021	0.025	0.006	0.002	0.235	0.490	-56.9	-46.3	-34.7	-28.2	-28.6	-28.6	-28.6	-6.0	-210
X-1	30	Imbr.Mol.	98.46	0.534	0.010	0.017	0.004	0.001	0.001	0.000	0.037	0.631	-62.8	-38.0	-25.2	-25.5	-20.4	-20.4	-20.4	-3.4	-187
X-2	190	Imbr.Mol.	98.01	0.678	0.135	0.036	0.015	0.014	0.003	0.000	0.022	0.567	-60.3	-43.3	-33.0	-26.9	-27.3	-27.3	-27.3	-3.0	-184
X-3	193	Imbr.Mol.	98.36	0.458	0.097	0.014	0.005	0.004	0.001	0.000	0.018	0.629	-61.5	-48.9	-36.0	-26.5	-25.5	-25.5	-25.5	-4.1	-180
X-4	-21	Imbr.Mol.	98.41	0.182	0.026	0.004	0.001	0.000	0.000	0.000	0.019	0.666	-63.2	-55.5	-38.5	-26.0	-21.5	-21.5	-21.5	-1.2	-182
Y-1	2026	Lower P.F.	98.07	0.432	0.215	0.078	0.024	0.028	0.008	0.002	0.152	0.427	-63.3	-51.8	-37.3	-33.1	-31.7	-31.7	-31.7	-10.9	-187
Z-1	544	H.F.	98.11	0.301	0.113	0.031	0.014	0.014	0.003	0.001	0.032	0.428	-59.7	-46.0	-30.0	-24.0	-27.0	-27.0	-27.0	-2.0	-204
AA-1	628	H.F.	98.17	0.281	0.103	0.033	0.011	0.013	0.002	0.001	0.074	0.523	-59.8	-47.7	-31.0	-23.0	-22.0	-22.0	-22.0	-0.4	-209
BB-1	463	H.F.	98.66	0.294	0.129	0.031	0.013	0.015	0.003	0.000	0.004	0.480	-60.8	-44.3	-29.7	-19.2	-20.8	-20.8	-20.8	-6.2	-201

Table 1 continued

Well	TVDS	Formation	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	i-C <sub>4</sub> H <sub>10</sub>	n-C <sub>4</sub> H <sub>10</sub>	i-C <sub>5</sub> H <sub>12</sub>	n-C <sub>5</sub> H <sub>12</sub>	n-C <sub>6</sub> H <sub>14</sub> +	CO <sub>2</sub>	N <sub>2</sub>	δ <sup>13</sup> C <sub>CH<sub>4</sub></sub>	δ <sup>13</sup> C <sub>C<sub>2</sub>H<sub>6</sub></sub>	δ <sup>13</sup> C <sub>C<sub>3</sub>H<sub>8</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>n-C<sub>5</sub>H<sub>12</sub></sub>	δ <sup>13</sup> C <sub>i-C<sub>4</sub>H<sub>10</sub></sub>	δ <sup>13</sup> C <sub>i-C<sub>5</sub>H<sub>12</sub></sub>	δ <sup>13</sup> C <sub>CO<sub>2</sub></sub>	δ <sup>2</sup> H <sub>CH<sub>4</sub></sub>
CC-1	335	H.F.	99.45	0.314	0.124	0.037	0.014	0.012	0.002	0.002	0.010	0.522	-60.3	-48.8	-31.0	-23.0	-23.0	-23.0	-23.0	-11.1	-201
DD-1	1454	Lower PF.	97.95	0.416	0.133	0.041	0.018	0.024	0.006	0.001	0.217	0.393	-57.0	-46.8	-32.4	-30.6	-30.6	-29.8	-29.8	-6.7	-186
DD-1	1278	Upper PF.	98.50	0.341	0.065	0.014	0.017	0.022	0.006	0.001	0.013	0.457	-59.0	-50.2	-35.1	-31.6	-31.6	-30.0	-30.0	-13.1	-189
EE-1	594	H.F.	98.47	0.291	0.101	0.039	0.012	0.014	0.003	0.000	0.079	0.448	-60.2	-43.6	-29.0	-21.0	-21.0	-22.0	-22.0	-2.0	-200
FF-1	1185	Upper PF.	98.50	0.441	0.109	0.040	0.016	0.032	0.006	0.024	0.039	0.597	-54.6	-49.1	-32.4	-27.0	-27.0	-28.0	-28.0	-13.0	-199
GG-1	493	H.F.	99.11	0.262	0.102	0.032	0.010	0.011	0.001	0.001	0.087	0.367	-60.3	-48.4	-30.9	-24.8	-24.8	-24.8	-24.8	-9.8	-198
GG-2	505	H.F.	98.76	0.325	0.131	0.032	0.014	0.013	0.003	0.003	0.052	0.655	-60.1	-43.8	-29.8	-24.0	-24.0	-23.0	-23.0	-9.3	-200
HH-1	1261	Upper PF.	98.09	0.326	0.109	0.035	0.016	0.020	0.005	0.001	0.117	0.513	-58.4	-46.1	-32.1	-26.0	-26.0	-26.7	-26.7	1.2	-205
II-1	419	H.F.	98.53	0.321	0.026	0.006	0.001	0.002	0.000	0.000	0.000	0.536	-62.0	-49.6	-27.0	-27.0	-27.0	-27.0	-27.0	-195	-195
II-2	623	H.F.	97.99	0.302	0.028	0.006	0.002	0.002	0.000	0.000	0.000	0.539	-60.7	-50.0	-27.3	-27.3	-27.3	-26.0	-26.0	-4.1	-206
JJ-1	1201	Upper PF.	97.81	0.390	0.118	0.035	0.019	0.024	0.006	0.002	0.239	0.463	-57.6	-49.7	-35.4	-26.0	-26.0	-26.4	-26.4	-5.5	-207
JJ-2	1181	Upper PF.	97.71	0.387	0.121	0.036	0.019	0.024	0.006	0.003	0.335	0.448	-57.7	-48.9	-35.1	-26.0	-26.0	-26.4	-26.4	-6.2	-207
JJ-3	1180	Upper PF.	97.58	0.391	0.115	0.034	0.019	0.025	0.006	0.000	0.456	0.463	-57.7	-49.7	-35.0	-24.5	-24.5	-24.0	-24.0	-4.6	-208
JJ-4	1186	Upper PF.	97.21	0.390	0.110	0.036	0.019	0.025	0.007	0.002	0.271	0.461	-57.7	-49.1	-35.3	-26.9	-26.9	-27.2	-27.2	-4.6	-208
KK-1	102	Upper PF.	97.47	0.023	0.000	0.000	0.000	0.000	0.000	0.000	0.002	1.806	-65.1	-47.9						-214	-214
KK-2	116	Upper PF.	97.93	0.022	0.000	0.000	0.000	0.000	0.000	0.000	0.002	2.106	-64.5	-42.3						-211	-211

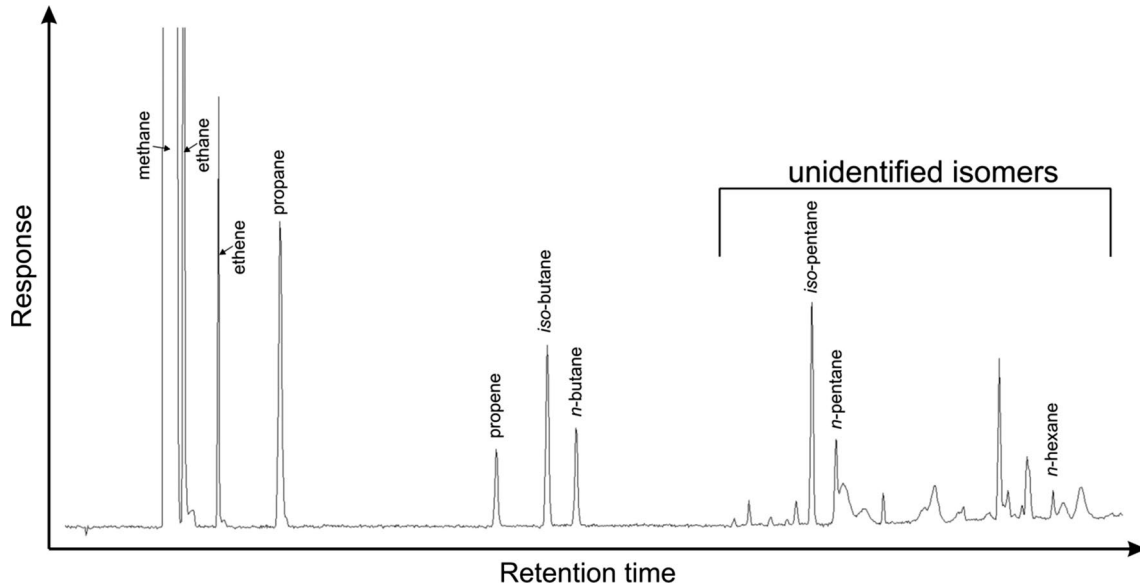
Due to confidentiality, in column "Well" true names are substituted by letters corresponding to fields (see Fig. 10) and digits to certain wells  
 TVDS true vertical depth subsea, H.F. Hall Formation, Upper P.F. Upper Puchkirchen Formation, Lower P.F. Lower Puchkirchen Formation, Imbr.Mol. Imbricated Molasse



thermogenic methane in the F–O fields. Apart from wet gas, some wells produce even liquid hydrocarbons (condensates), another proof for the presence of thermogenic input (Pytlak et al. 2014; Wagner 1998). Traces of higher hydrocarbons are also visible in total gas composition (Fig. 4). Interestingly, gas dryness in Puchkirchen and Hall

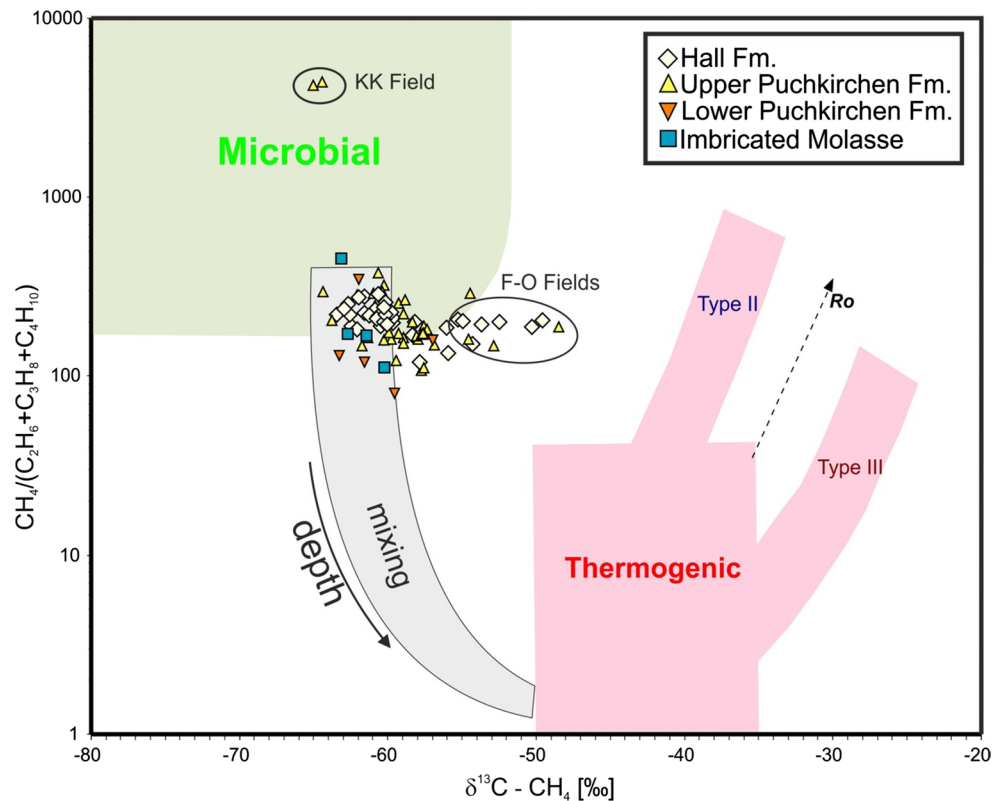
reservoirs with liquid hydrocarbons is often relative high (>200; Fig. 5).

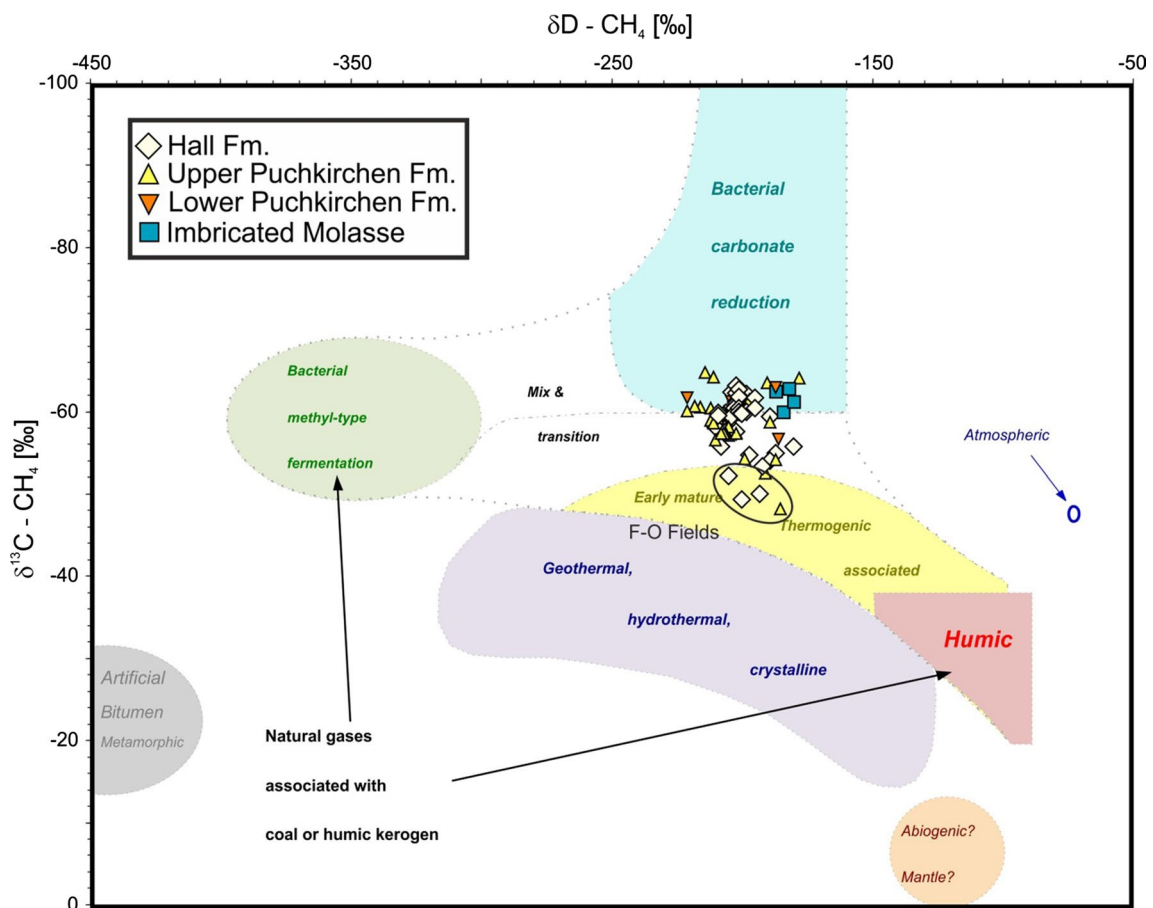
Ethane in many samples is strongly depleted in <sup>13</sup>C (Fig. 7). This suggests that most of the detected ethane is microbial in origin. Actually, ethane and propane can be generated by microbial activity, although their contribution



**Fig. 4** Flame ionization detector (FID) chromatogram of gas sample from gas field F. For location see Fig. 2

**Fig. 5** Genetic characterization of oil-associated gas from the Austrian part of the Alpine Foreland Basin (discrimination diagram adopted from Bernard et al. 1978; Whiticar and Suess 1990). Note that y axis of original plot is expressed as  $CH_4/(C_2H_6 + C_3H_8)$ . Mixing trend with depth is determined by results of mud gas measurements from well-penetrating Imbricated Molasse (unpublished industrial data, for location see Fig. 2)





**Fig. 6** Genetic characterization of gas from Oligocene–Miocene reservoirs in the Austrian part of the Alpine Foreland Basin (plot modified after Whiticar et al. 1986)

in total gas composition is typically very low (Hinrichs et al. 2006; Oremland et al. 1988; Rice and Claypool 1981).

### Origin of thermogenic hydrocarbons

Figure 8a shows that gas dryness decreases with reservoir depth implying a downward increase in thermogenic hydrocarbons (see also Reischenbacher and Sachsenhofer 2011). Potentially these hydrocarbons may have been generated locally in pelitic rocks within the studied succession (or below) or may have been derived from oil-bearing deeper horizons (e.g., Eocene reservoir rocks).

The Cenozoic succession includes potential source rocks. However, these are immature north of the Alpine front (e.g., Gusterhuber et al. 2012, 2013). This is not only valid for Lower Oligocene source rocks, but also for “fish shales” near the top of the Upper Puchkirchen Formation containing up to 3.5% TOC (Belaed 2007).

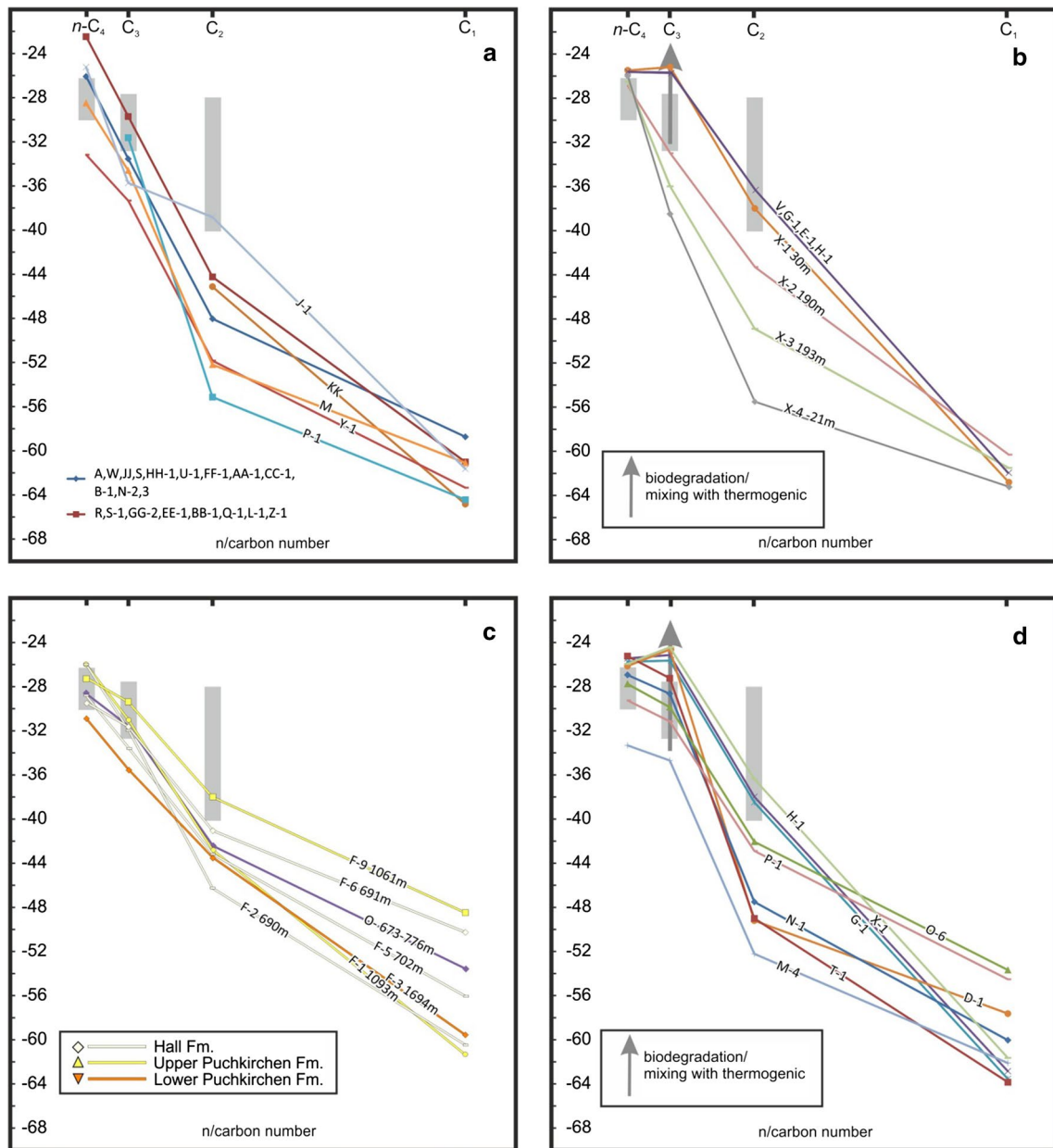
Therefore, it is likely that the thermogenic hydrocarbons migrated upward from Eocene carrier beds and reservoir rocks (e.g., Wagner 1998). Although commercial

oil deposits in the western part of the study area are missing, industry data show that wet gas and oil stains within Eocene horizons have been detected in the entire study area.

The plot of  $\delta^{13}\text{C}$  of ethane versus  $\delta^{13}\text{C}$  of propane (Berner and Faber 1996; Fig. 9) is often used to determine the maturity of gases. The data from Oligocene–Miocene reservoirs are generally shifted toward light  $\delta^{13}\text{C}$  values of ethane (Fig. 9). This may indicate a contribution of microbial ethane. Moreover, propane may be enriched in  $^{13}\text{C}$  due to biodegradation (see “Gas biodegradation and secondary microbial methane” section). Nevertheless, propane and *n*-butane isotope ratios are similar to those measured in oil-associated gas from Mesozoic and Eocene reservoirs (Pytlak et al. 2016; Fig. 7), suggesting oil-window maturity.

### Controls of mixing of thermogenic and microbial gas

Whereas the microbial component of the gas has been generated locally within the Upper Oligocene to Lower Miocene succession (Schulz and van Berk 2009), the



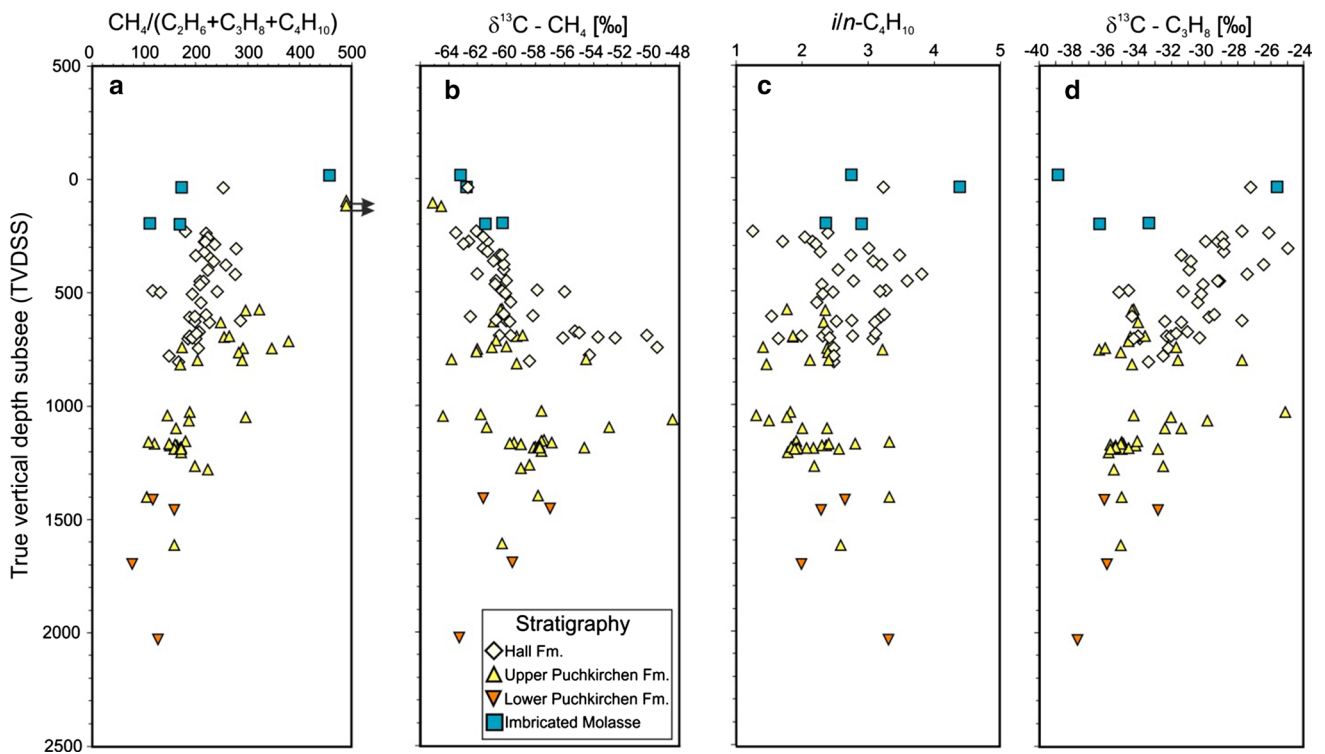
**Fig. 7** Natural gas plots (Chung et al. 1988) showing the carbon isotopic compositions of individual hydrocarbons as a function of the carbon number. Depths are true vertical depth subsea (TVDSS)

expressed in meters. Gray bars indicate isotopic signature of hydrocarbons associated with oil deposits in the Austrian part of Alpine Foreland Basin (Pytlak et al. 2016)

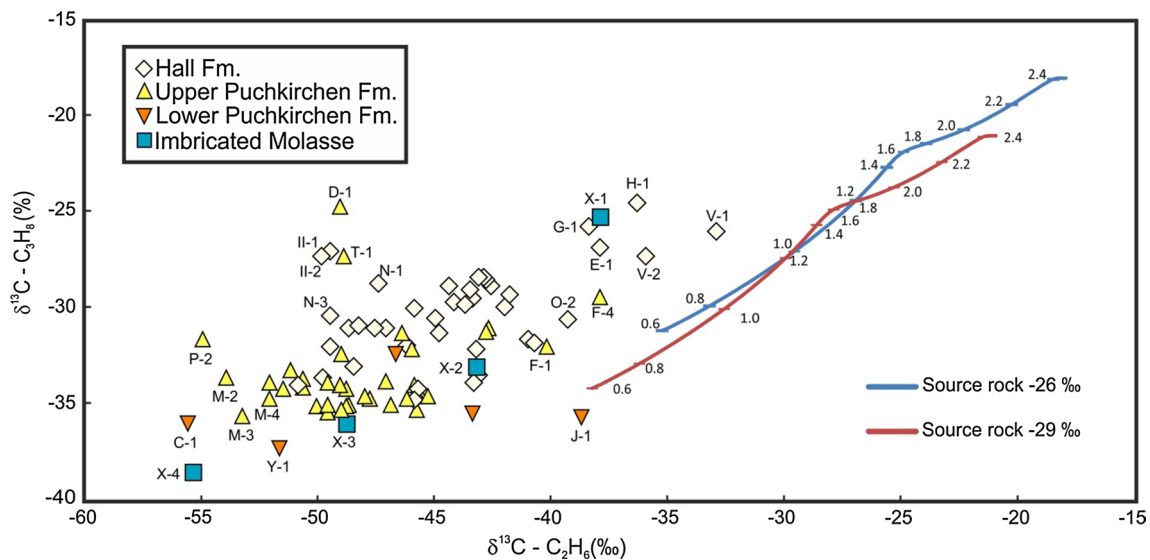
thermogenic fraction migrated vertically upward from Mesozoic and Eocene carrier beds and/or reservoirs. Migration may have occurred (1) along relative permeable faults or (2) within low-permeability rocks along subseismic fractures.

1. Samples G-1, E-1, H-1 and gas from a V field produce methane from Hall reservoirs in the eastern part of the study area.  $\delta^{13}\text{C}$  values of methane in these fields are

characteristic for microbial origin, while those of  $\text{C}_2$  to  $\text{C}_4$  hydrocarbons are characteristic for thermogenic, oil-associated gas (Fig. 7b). The producing Hall horizons in the above fields are located above or close to oil deposits in Eocene strata. Faults were active in the Alpine Foreland Basin during different times. In some areas fault activity stopped already during deposition of the Lower Puchkirchen Formation, whereas in others it continued till Eggenburgian time (Fig. 10). These



**Fig. 8** Depth distribution of selected geochemical parameters.  $i/n$ - $C_4H_{10}$  ratios  $>1$  indicate that all gases are biodegraded. However, dryness and  $\delta^{13}C$ - $C_3H_8$  trends indicate that biodegradation decreases with depth



**Fig. 9** Relative gas maturity base  $\delta^{13}C$  distribution in ethane and propane (plot modified after Berner and Faber 1996)

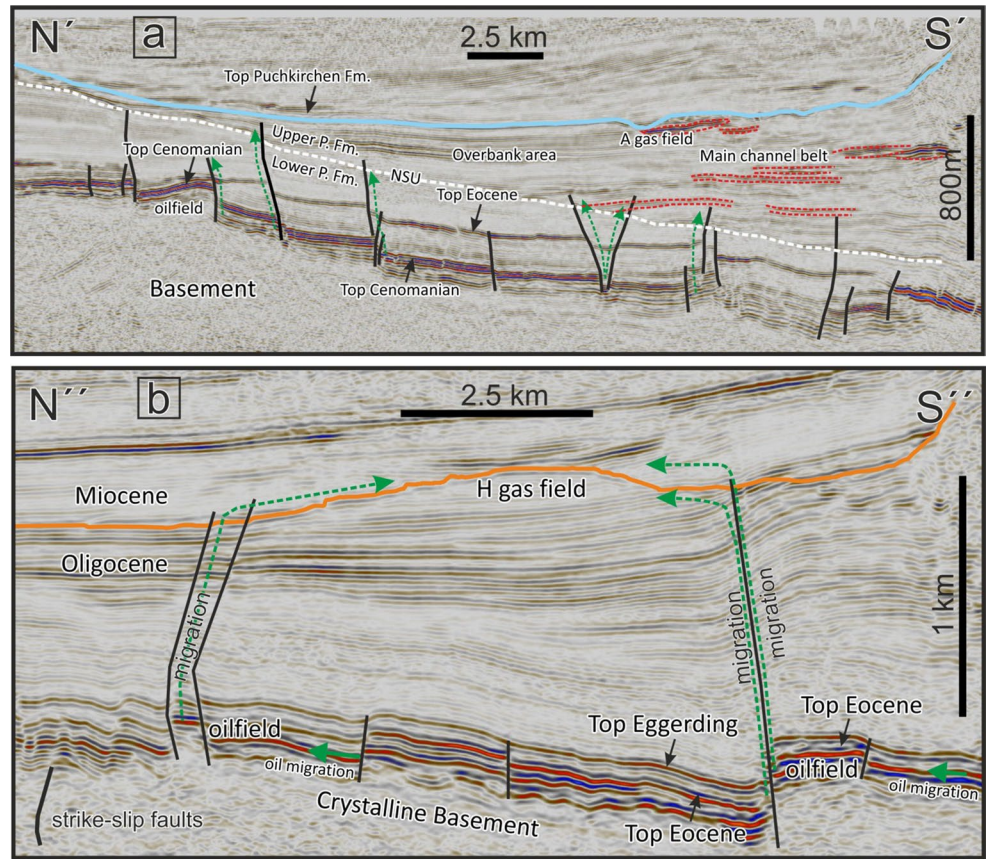
long-lived faults provided potential migration pathways connecting Eocene reservoirs and the Hall Formation (Wagner 1998).

2. In areas without distinct faults, migration may have occurred through low-permeability caprocks along subseismic fractures, silty intervals and/or by diffu-

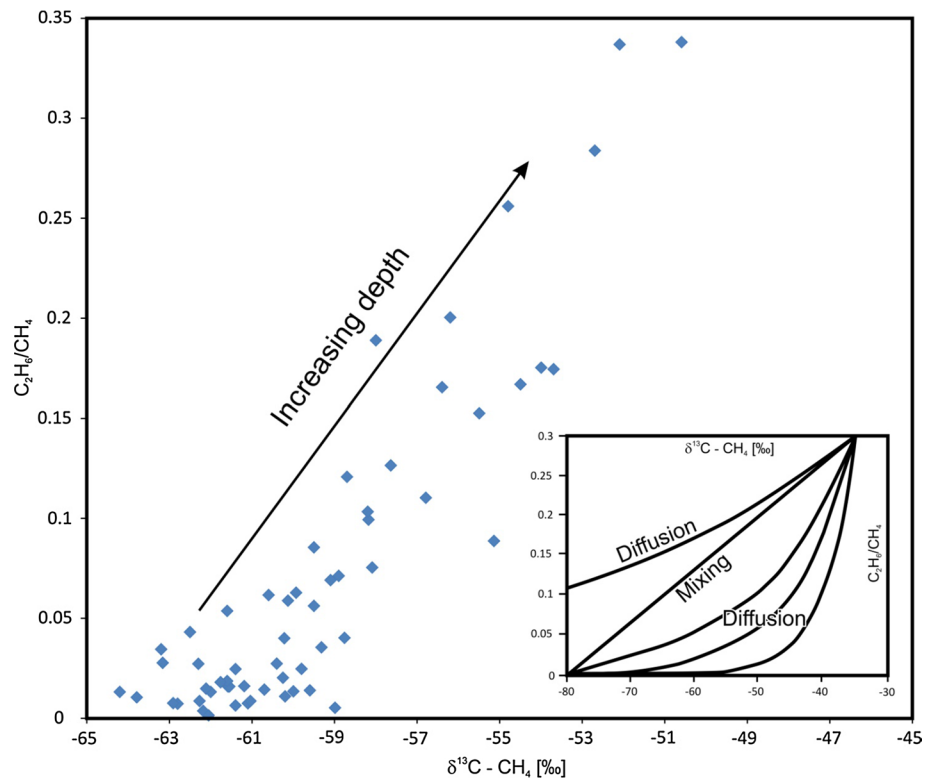
sion. It is likely that upward migration may be facilitated in areas where lower Oligocene fine-grained rocks are eroded and along the Puchkirchen Channel system, where a high percentage of permeable clastic rocks have been deposited. Within this context, it is remarkable to observe that the F–O fields containing

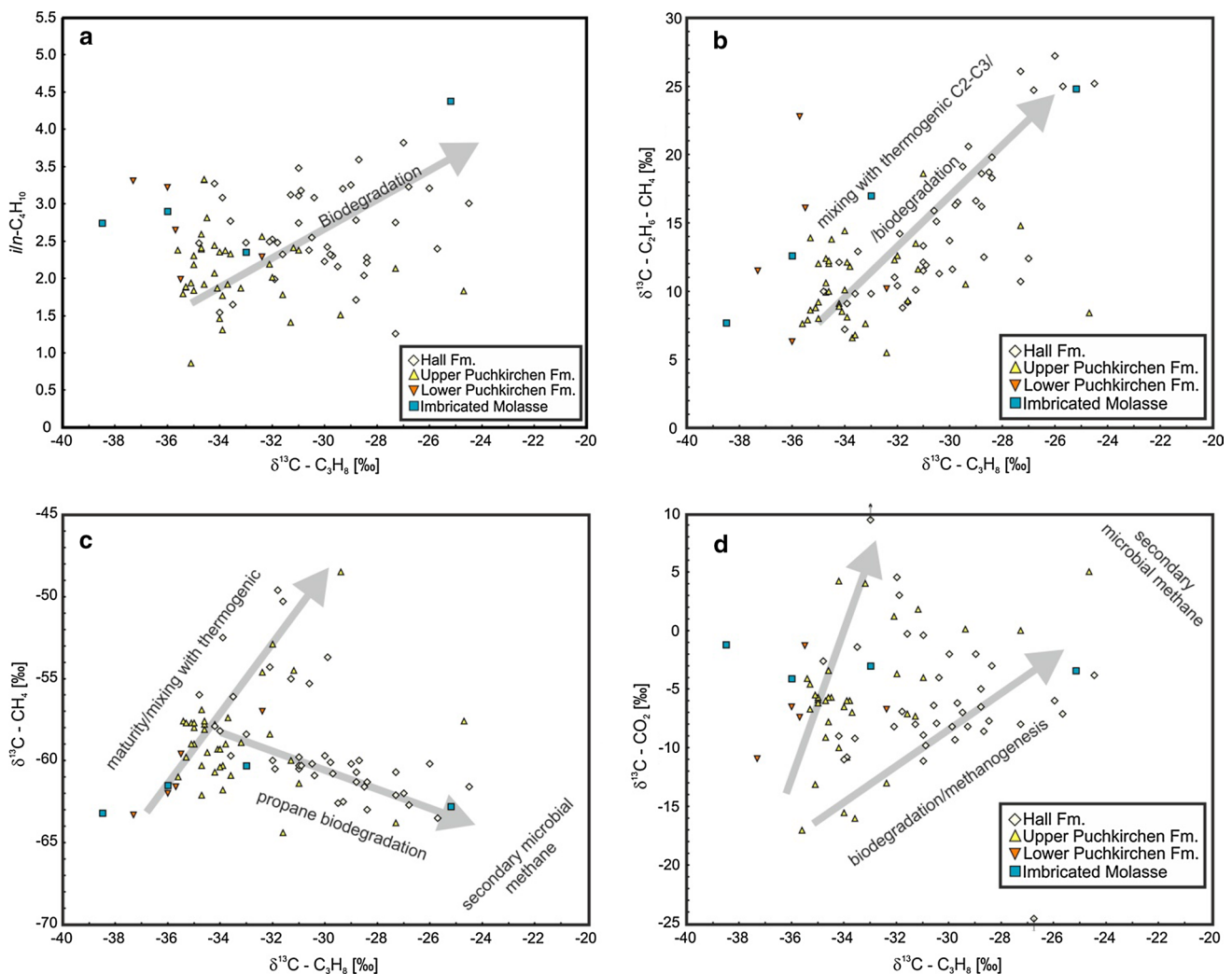


**Fig. 10** Seismic examples of the Alpine Foreland Basin (for location see Fig. 2). **a** Across gas field A. Location of Puchkirchen Channel belt and adjacent overbank deposits (both are the main gas reservoirs) are indicated with *dashed red lines*. Northern Slope Unconformity (NSU, Masalimova et al. 2015) separates Lower Puchkirchen Formation (Lower P. Fm.) from Upper Puchkirchen Formation (Upper P. Fm.). The NSU is cross-cut by faults active during deposition of Upper Puchkirchen Formation providing potential migration pathways for thermogenic hydrocarbons (*dashed green lines*). **b** Across gas field H and oil fields. Majority of oil and thermogenic gas in the Alpine Foreland Basin are trapped in Cenomanian and Eocene reservoirs in fault-related structures



**Fig. 11** Cross-plot of  $C_1/C_2$  molecular ratio versus  $\delta^{13}C$  of methane (after Prinzhofer and Pernaton 1997) of mudgas from a well near the southern margin of the Alpine Foreland Basin. Inset presents theoretical mixing and diffusion trends. Diffusion trends depend on ratios of the diffusive permeabilities for ethane over methane





**Fig. 12** Correlation of molecular and isotopic composition. Gray arrows indicate possible selective components removal and mixing with thermogenic hydrocarbons

isotopically methane enriched in  $^{13}\text{C}$  ( $-48$  to  $-61.4$ ‰) testifying to a high amount of thermogenic gas are located both along the axis of the Puchkirchen Channel and above an area where more than 60 m of lower Oligocene fine-grained rocks have been removed by submarine erosion (Sachsenhofer and Schulz 2006). In contrast, the KK field hosting no thermogenic hydrocarbons, but pure microbial gas is located far north of the Puchkirchen Channel and in an area with thick pelitic Lower Oligocene deposits.

Migration through low-permeability caprocks may involve diffusion. Diffusion processes lead to molecular and isotopic fractionation due to differences in the effective diffusion coefficients (e.g., Krooss et al. 1988; Leythaeuser et al. 1983; Xia and Tang 2012; Zhang and Krooss 2001).

Prinzhofer and Pernaton (1997) proposed a plot of the  $\text{C}_2/\text{C}_1$  ratio versus  $\delta^{13}\text{C}$  of methane (Fig. 11) to distinguish diffusive fractionation from simple mixing of two end-member gases. Because the Prinzhofer plot may be used for gases in seal rocks, but not for reservoir gases (Zhang and Krooss 2001) and because the thermogenic end-member gas in the case of the Alpine Foreland Basin is a gas mixture itself (Pytlak et al. 2016), we do not plot reservoir gas data, but mud gas data from a recently drilled well, which penetrated shaly units, several hundred meters thick. The mud gas data show a strong linear relationship between the  $\text{C}_2/\text{C}_1$  ratio and  $\delta^{13}\text{C}$  of methane (in the plot with linear axis) suggesting that mixing processes are dominant, whereas fractionation associated with diffusion does not appear to have influenced gas compositions significantly. Obviously, diffusive fractionation cannot be completely



ruled out, but its effect on the governing molecular and isotopic gas composition may have been overprinted by other processes (mixing, alteration; see below).

### Gas biodegradation and secondary microbial methane

During microbial alteration of gaseous hydrocarbons propane is primarily degraded (Head et al. 2003; Kniemeyer et al. 2007). In addition, straight chain alkanes are preferentially removed compared to branched ones (James and Burns 1984; Palmer 1993) resulting in increasing *i/n*-C<sub>4</sub> ratios. Although *i/n*-C<sub>4</sub> ratio can depend on maturity (Alexander et al. 1981), those of non-degraded gases are usually below 0.8 (Huang and Larter 2014). In most cases microbial activity leads to enrichment in <sup>13</sup>C isotope of remaining molecules (e.g., C<sub>3</sub>, *n*-C<sub>4</sub>).

To visualize gas alteration trends, *i/n*-C<sub>4</sub> ratios and stable carbon isotopic composition of propane are plotted versus depth in Fig. 8c, d. *i/n*-C<sub>4</sub> ratios significantly above 0.8 indicate that all gases are biodegraded. *i/n*-C<sub>4</sub> ratios show a weak, <sup>13</sup>C ratios of propane a strong downward decreasing trend providing evidence that the effect of in-reservoir biodegradation decreases downward with increasing formation temperature. Although microbes may be active at high temperatures, it is commonly accepted that microbial activity relevant for petroleum formation and degradation is restricted to temperatures below 80 °C (Head et al. 2003 and references therein). This observation agrees with the fact that in the study area a formation temperature of 80 °C is reached between 1500 and 2000 m depth subsea (Kamyar 2000). However, it is important to note that regional uplift after Late Miocene maximum burial resulted in cooling of about 20 °C (Gusterhuber et al. 2012). This argues for biodegradation, at least in deep reservoirs, before or after maximum burial.

The natural gas plot (Chung et al. 1988) shows that a significant number of samples show deviations from typical <sup>13</sup>C distribution patterns ( $\delta^{13}\text{C}_{\text{C}_2\text{H}_6} < \delta^{13}\text{C}_{\text{C}_3\text{H}_8} < \delta^{13}\text{C}_{\text{C}_4\text{H}_{10}}$ ; Schoell 1984; Tang et al. 2000) (Fig. 7b, d). This is caused mainly by an enrichment of <sup>13</sup>C in propane and reflects the selective microbial removal of this compound.

Some samples contain ethene and propene (Fig. 8), which are considered instable over geological timescales. Although microbial processes yielding unsaturated hydrocarbons are poorly understood, their presence may indicate recent bacterial activity (Whiticar 1994). Microbial activity is also confirmed by the presence of metabolites in reservoir rocks (Gruner, pers. comm.). Drill bit metamorphism as source for unsaturated hydrocarbons in reservoirs, which are in production since several years, can be ruled out. Whereas (ongoing) biodegradation is well confirmed, only a weak correlation is observed in the cross-plot of *i/n*-C<sub>4</sub> versus <sup>13</sup>C of propane (Fig. 12a).

A general positive relation exists between the difference of <sup>13</sup>C of methane and ethane and the <sup>13</sup>C of propane (Fig. 12b). This may result from either gas mixing or biodegradation. A cross-plot of <sup>13</sup>C of methane versus <sup>13</sup>C of propane shows two separate trends indicative for mixing and biodegradation of propane, respectively (Fig. 12c).

Recently, biodegradation of hydrocarbons followed by methanogenesis has been recognized as a significant source of (secondary microbial) gas (e.g., Head et al. 2003; Jones et al. 2008; Milkov 2010; Zengler et al. 1999). The isotopic signature of secondary microbial methane depends on different factors, including the methanogenic community, reaction pathways, progress of alteration, isotopic signature of the substrate and reservoir temperature (Brown 2011; Jones et al. 2008). Frequently, secondary microbial methane is accompanied by CO<sub>2</sub> enriched in the <sup>13</sup>C isotope (up to 20‰) (Milkov 2011 and references therein). Methane depleted in <sup>13</sup>C (−65‰ or lower) is produced when CO<sub>2</sub> reduction commences, while CO<sub>2</sub> is moderately enriched in <sup>13</sup>C isotope (Jones et al. 2008). However, as CO<sub>2</sub> reduction progresses, carbon in both, CO<sub>2</sub> and methane, become heavier. CO<sub>2</sub> concentrations in total gas volume are not necessarily high, since CO<sub>2</sub> can be transformed into calcite cement (Dimitrakopoulos and Muehlenbachs 1987; Wiggins et al. 1993). Actually, Grundtner et al. (2014) could show that organic matter-derived CO<sub>2</sub> triggering calcite precipitation plays an important role in the diagenetic history of the Alpine Foreland Basin. In the present study measured <sup>13</sup>C of CO<sub>2</sub> reaches values up to 18‰ with an average value of −5.9‰ (Table 1; Fig. 12d). Moreover, a general positive relation between <sup>13</sup>C of CO<sub>2</sub> and C<sub>3</sub>H<sub>8</sub> points to methanogenesis (Table 1; Fig. 12d). Therefore, biodegradation and formation of secondary microbial gas are considered at least as a minor additional source of methane.

Generation of secondary microbial methane results in gas drying. This may explain the relative high dryness values of gas samples with high <sup>13</sup>C ratios.

### Conclusions

Gas trapped in Oligocene–Miocene reservoir rocks in the Alpine Foreland Basin has traditionally been considered (primary) microbial in origin. A detailed investigation of samples from all producing gas fields provides a more differentiated picture, which contributes significantly to the understanding of the petroleum system.

- Dryness and the most negative <sup>13</sup>C values of methane suggest a pure primary microbial origin of gas in Oligocene–Miocene reservoir horizons in the KK field, confirming the traditional model.

- In all others samples heavier hydrocarbons are present. The amount of  $C_{2+}$  hydrocarbons increases with reservoir depth, indicating a downward increasing contribution of thermogenic hydrocarbons. Varying contributions of thermogenic hydrocarbons are also reflected by strongly varying  $\delta^{13}C$  values of methane. Based on  $\delta^{13}C$  values of methane, gas from the F–O field in the western part of the study area contains the highest contribution of thermogenic hydrocarbons.
- The thermogenic hydrocarbons display an oil-window maturity and are derived from the deeper Mesozoic and Eocene oil-bearing reservoir and carrier beds. Upward migration along faults is likely (e.g., fields in the eastern part of the study area). Migration also occurred through low-permeability seal rocks (e.g., along subseismic fractures). Migration may have been facilitated in areas where lower Oligocene pelitic rocks are eroded and along the Puchkirchen Channel system, where a high percentage of permeable clastic rocks have been deposited. An effect of diffusion on molecular and isotopic composition of the gas cannot be observed.
- All gas samples in Oligocene–Miocene reservoirs are biodegraded, but the degree of biodegradation decreases downward. Biodegradation results in an increase in the  $i/nC_4$  ratio, a selective removal of propane, an increase in  $\delta^{13}C$  of propane and gas drying.
- Biodegradation is accompanied with the generation of secondary microbial methane (and  $^{13}C$  enriched  $CO_2$ ) resulting in further drying of the gas. Hence, gas dryness is high even in samples with high contributions of thermogenic hydrocarbons.
- The presence of alkenes suggests that biodegradation is an ongoing process.

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